

Reaction of *d*-Pseudoephedrine Hydrochloride with Anisaldehyde in the Presence of Sodium Cyanide

A. B. Bukeeva, A. M. Gazaliev, D. M. Turdybekov,
O. A. Nurkenov, K. M. Turdybekov, and G. G. Baikenova

Institute of Organic Synthesis and Coal Chemistry,

Ministry of Education and Science of Kazakhstan, Karaganda, Kazakhstan

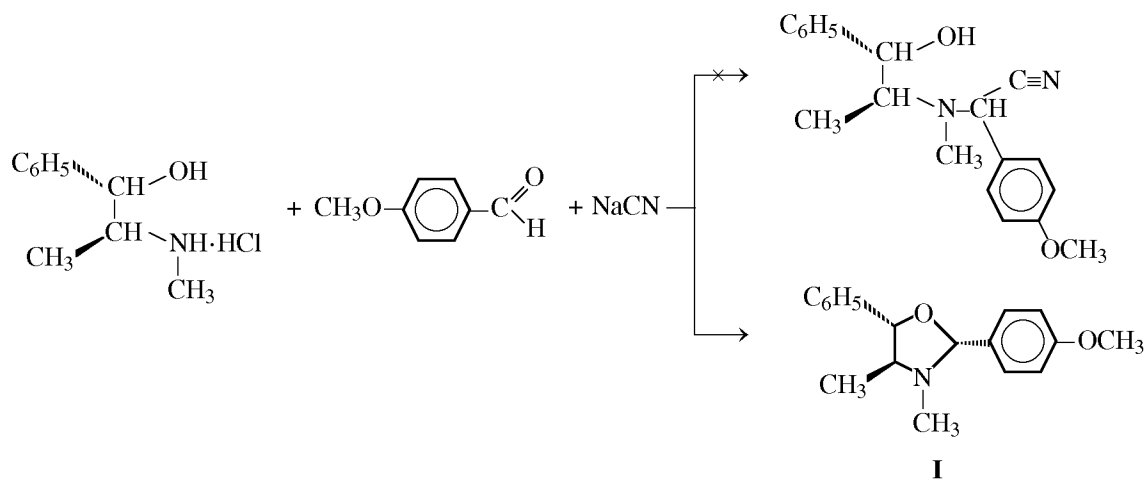
Institute of Phytochemistry, Ministry of Education and Science of Kazakhstan, Karaganda, Kazakhstan

Received May 21, 2001

Abstract—2-(*p*-Methoxyphenyl)-3,4-dimethyl-5-phenyl-1,3-oxadiazolidine was synthesized, and its steric structure was studied by X-ray diffraction.

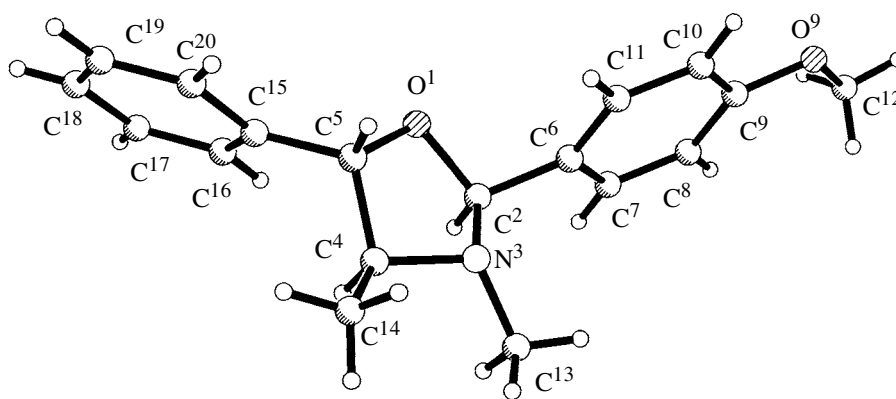
As reported earlier [1], ephedrine alkaloids (*l*-ephedrine and *d*-pseudoephedrine) react with symmetrical cyanhydrines to give corresponding nitriles. The reaction of *l*-ephedrine with aromatic aldehydes in the presence of sodium cyanide gives rise to α -aminonitriles [2, 3]. However, reacting *d*-pseudoephedrine with anisaldehyde in similar conditions we found that the

synthesis leads to a heterocyclic compound, 2-(*p*-methoxyphenyl)-3,4-dimethyl-5-phenyl-1,3-oxadiazolidine (**I**), rather than the expected α -aminonitrile. Apparently, this result is explained by the fact that *d*-pseudoephedrine hydrochloride in the presence of sodium cyanide transforms into a base which condenses with the aldehyde to give the heterocyclic compound.



The steric structure of compound **I** was established by X-ray diffraction. The resulting structure is shown in the figure. The bond lengths and bond angles are close to normal [4]. The oxadiazolidine ring in **I** takes a slightly distorted 3-*envelope* conformation (ΔC_s^3 8.4°) (the endocyclic torsion angles $O^1C^2N^3C^4$, $C^2N^3C^4C^5$, $N^3C^4C^5O^1$, $C^4C^5O^1C^2$, and $C^5O^1C^2N^3$ are 42.0, –46.6, 33.2, –7.5, and –21.5°, respectively). The N^3 atom

deviates from the plane defined by the other ring atoms by 0.667 Å, and the O^1 , C^2 , C^4 , and C^5 atoms are coplanar within ± 0.033 Å. In the 3-*envelope* conformation, the methyl groups on N^3 and C^4 and the methoxyphenyl group are equatorial (the torsion angles $C^{13}N^3C^4C^5$, $C^{14}C^4N^3C^2$, and $C^6C^2N^3C^4$ are –167.6, –168.6, and 162.0°, respectively), while the phenyl group is pseudoequatorial (the torsion angle

Molecular structure of compound **I**.

$C^{15}C^5C^4N^3$ is 154.9°). The 3-*envelope* conformation is also characteristic of (2*S*,4*S*,5*S*)-3,4-dimethyl-5-phenyl-2-(phenylethynyl)-1,3-oxazolidine [5].

Since the *l*-ephedrine and *d*-pseudoephedrine derivatives bear substituents on N^3 , C^4 , and C^5 , one more favorable conformation of the ring is 4-*envelope*, where the methyl group on C^4 is equatorial, and the substituents on the other two mentioned atoms are pseudoequatorial. The 4-*envelope* conformation is most common in pseudoephedrine oxazolidine derivatives, specifically, 2-(methoxycarbonyl)-3,4-dimethyl-5-phenyl-1,3-oxazolidine [6].

EXPERIMENTAL

The IR spectrum was measured on a UR-20 instrument (KBr). The 1H NMR spectrum was obtained on a Tesla BS-587 spectrometer (80 MHz) in C_6D_6 relative to internal HMDS. The melting point was measured on a Boëtuis hot stage.

X-ray diffraction experiment. The unit cell parameters and the intensities of 1702 unique reflections of crystal **I** were measured at $20^\circ C$ on a Bruker-P4 automatic four-circle diffractometer with monochromatic λMoK_α radiation ($\infty/2\infty$ scanning, $2\theta \leq 60^\circ$). Orthorhombic crystals, a 6.156(1), b 14.211(3), c

Atomic coordinates ($\times 10^4$; for H atoms, $\times 10^3$) in molecule **I**

Atom	<i>x</i>	<i>y</i>	<i>z</i>	Atom	<i>x</i>	<i>y</i>	<i>z</i>
O ¹	2542(4)	3378(2)	1401(2)	H ³	561(7)	291(3)	37(3)
O ²	5263(6)	1480(3)	5313(2)	H ⁴	684(6)	115(3)	14(2)
N ¹	2325(4)	1668(2)	1656(2)	H ⁵	515(7)	-23(3)	80(2)
C ²	3259(6)	2516(2)	1256(2)	H ⁷	209(6)	-78(3)	164(2)
C ³	4997(7)	2296(3)	700(2)	H ^{8a}	-108(6)	-56(3)	246(2)
C ⁴	5644(7)	1335(3)	556(2)	H ^{8b}	-109(6)	28(3)	155(2)
C ⁵	4648(6)	510(3)	969(2)	H ⁹	-209(6)	119(3)	291(2)
C ⁶	3012(5)	690(2)	1517(2)	H ^{10a}	-62(6)	220(3)	183(2)
C ⁷	1944(6)	-170(2)	1995(2)	H ^{10b}	94(6)	250(3)	258(2)
C ⁸	-394(6)	65(3)	2148(3)	H ^{11a}	71(6)	146(3)	392(2)
C ⁹	-435(5)	1020(3)	2700(2)	H ^{11b}	-12(5)	31(2)	388(2)
C ¹⁰	472(6)	1928(3)	2207(3)	H ^{13a}	463(5)	-51(2)	276(2)
C ¹¹	742(6)	831(3)	3540(2)	H ^{13b}	234(5)	-98(2)	312(2)
N ¹²	2962(4)	562(2)	3374(2)	H ^{14a}	353(6)	2(2)	458(2)
C ¹³	3069(6)	-359(2)	2847(2)	H ^{14b}	565(6)	-1(2)	409(2)
C ¹⁴	4196(7)	452(3)	4157(2)	H ^{16a}	499(8)	228(3)	347(3)
C ¹⁵	4852(7)	1448(3)	4558(2)	H ^{16b}	599(8)	280(3)	417(3)
C ¹⁶	5029(10)	2385(3)	4017(3)	H ^{16c}	359(6)	282(5)	402(4)

18.878(1) Å; V 1651.5(2) Å³, d_{calc} 1.14 g/cm³, Z 4, C₁₈H₂₁NO₂, space group $P2_12_12_1$.

In calculations we used 942 reflections with $I \geq 2\sigma(I)$. The structure was solved by the direct method and refined by full-matrix least squares in the anisotropic approximation for non-hydrogen atoms. Hydrogen atoms were fixed geometrically and were not refined. Weight parameter 0.1725. The structure was refined to R 0.0705 and R_w 0.2332. All calculations were performed on an RS/AT-200 computer using the SHELXL 97 program package. The coordinates of non-hydrogen atoms are given in the table.

(2*S*,4*S*,5*S*)-2-(*p*-Methoxyphenyl)-3,4-dimethyl-5-phenyl-1,3-oxazolidine (I). To a mixture of 2.8 g of *d*-pseudoephedrine hydrochloride in 5 ml of acetonitrile we added with stirring at room temperature 3.59 g of *p*-methoxybenzaldehyde and then, slowly, a solution of 0.68 g of sodium cyanide in 2 ml of water. The mixture was stirred at room temperature for 5 h. The reaction products were extracted with ethyl acetate, the extracts were dried with MgSO₄, the solvent was removed by distillation, and the residue was passed through a silica gel column, eluent ethyl acetate–hexane (15:85). Yield 1.68 g (42.8%), mp 62–63°C. ¹H NMR spectrum, δ , ppm: 1.15 d (3H, CH₃CH, J_{HH} 5.8 Hz), 2.15 s (3H, CH₃N), 2.52 m (1H,

CHN), 3.82 s (3H, CH₃O), 5.10 s (1H, NCHO), 4.80 d (1H, CHO, J_{HH} 6.6 Hz), 7.51 m (5H, C₆H₅). Found, %: C 76.01; H 7.38; N 4.91. C₁₈H₂₁NO₂. Calculated, %: C 76.32; H 7.42; N 4.95.

REFERENCES

1. Gazaliev, A.M., Nurkenov, O.A., and Zhurinov, M.Zh., *Khimiya α -aminonitrilov* (Chemistry of α -Aminonitriles), Almaty: Gylym, 1998.
2. Nurkenov, O.A., Gazaliev, A.M., Bukeeva, A.B., Baiknova, G.G., and Zhurinov, M.Zh., *Zh. Obshch. Khim.*, 2001, vol. 71, no. 1, p. 167.
3. Chang, C.J., Fang, J.M., Lee, C.H., and Wang, Y., *J. Chem. Soc., Perkin Trans. 1*, 1994, vol. 207, no. 24, p. 3587.
4. Allen, F.H., Kennard, O., Watson, D.G., Brammer, L., Orpen, A.G., and Taylor, R., *J. Chem. Soc., Perkin Trans. 2*, 1987, no. 1, p. 1.
5. Nurkenov, O.A., Markova, I.V., Shalbaeva, A.B., Turdybekov, K.M., and Gazaliev, A.M., *Zh. Obshch. Khim.*, 1999, vol. 69, no. 4, p. 679.
6. Bellan, J., Rossi, J.C., Cherean, N., Rogues, R., Germain, G., and Declereg, J.P., *Acta Crystallogr., Sect. C*, 1978, vol. 34, no. 10, p. 1648.